

The investigations were supported by grants from the Polish Academy of Sciences (Nos. MR-I.12 and C-1.1).

References

- ANET, F. A. L. & BOURN, A. J. R. (1965). *J. Am. Chem. Soc.* **87**, 5250–5251.
- APSIMON, J. W., DEMARCO, P. V., MATHIESON, D. W., CRAIG, W. G., KARIM, A., SAUNDERS, L. & WHALLEY, W. B. (1970). *Tetrahedron*, **26**, 119–146.
- BOTHNER-BY, A. A. & POPE, J. A. (1965). *Annu. Rev. Phys. Chem.* **16**, 43–64.
- DORMAN, D. E. & BOVEY, F. A. (1973a). *J. Org. Chem.* **38**, 1719–1722.
- DORMAN, D. E. & BOVEY, F. A. (1973b). *J. Org. Chem.* **38**, 2379–2383.
- HELGESON, R. C., LAUER, M. & CRAM, D. J. (1983). *J. Chem. Soc. Chem. Commun.* pp. 101–103.
- International Tables for X-ray Crystallography* (1974). Vol. IV. Birmingham: Kynoch Press.
- JOHNSON, C. K. (1965). *ORTEP*. Report ORNL-3794. Oak Ridge National Laboratory, Tennessee.
- KOZERSKI, L. (1981). *J. Mol. Struct.* **75**, 95–99.
- LA PLANCHE, L. A. & ROGERS, M. T. (1963). *J. Am. Chem. Soc.* **85**, 3728–3730.
- LEVY, G. C. & NELSON, G. L. (1972). *J. Am. Chem. Soc.* **94**, 4897–4901.
- LEWIN, A. H. & FRUCHT, M. (1975). *Org. Magn. Reson.* **7**, 206–225.
- LINDEMAN, S. V., SHKLOVER, V. E. & STRUCHKOV, YU. T. (1981). *Cryst. Struct. Commun.* **10**, 1173–1179.
- MCFARLANE, W. (1970). *J. Chem. Soc. Chem. Commun.* pp. 418–419.
- MACNICOL, D. D., MCKENDRICK, J. J. & WILSON, D. R. (1978). *Chem. Soc. Rev.* **7**, 65–87.
- PICCINI-LEOPARDI, C. & REISSE, J. (1981). *J. Magn. Reson.* **42**, 60–72.
- PONNUSWAMY, M. N. & PARTHASARATHY, S. (1981). *Cryst. Struct. Commun.* **10**, 1203–1209.
- SHELDRIK, G. M. (1976). *SHELX76*. Program for crystal structure determination. Univ. of Cambridge, England.
- STEWART, J. M., KUNDELL, F. A. & BALDWIN, J. C. (1970). The *XRAY70* system. Computer Science Center, Univ. of Maryland, College Park, Maryland.
- STEWART, W. E. & SIDDALL, T. H. III (1970). *Chem. Rev.* **70**, 517–551.
- URBAŃCZYK-LIPKOWSKA, Z., KRAJEWSKI, J. W., GLUZIŃSKI, P., ANDRETTI, G. D. & BOCELLI, G. (1982). *Cryst. Struct. Commun.* **11**, 1537–1542.
- URBAŃCZYK-LIPKOWSKA, Z., KRAJEWSKI, J. W., GLUZIŃSKI, P., KOZERSKI, L. & CZUGLER, M. (1982). *Cryst. Struct. Commun.* **11**, 1531–1536.
- VOELTER, W. & OSTER, O. (1973). *Org. Magn. Reson.* **5**, 547–548.

Acta Cryst. (1984). **C40**, 1753–1756

10-(2-Methoxyphenyl)pyrido[3,2-*b*][1,4]benzothiazine, C₁₈H₁₄N₂O₅

BY PATRICE DE MEESTER AND SHIRLEY S. C. CHU*

School of Engineering and Applied Science, Southern Methodist University, Dallas, Texas 75275, USA

AND MISA V. JOVANOVIĆ AND EDWARD R. BIEHL

Department of Chemistry, Southern Methodist University, Dallas, Texas 75275, USA

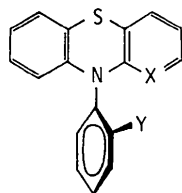
(Received 15 March 1984; accepted 5 June 1984)

Abstract. $M_r = 306.38$, monoclinic, $P2_1/c$, $a = 11.859$ (1), $b = 8.121$ (1), $c = 16.016$ (1) Å, $\beta = 101.59$ (1)°, $V = 1511.0$ (3) Å³, $Z = 4$, $D_x = 1.347$ g cm⁻³, $\lambda(\text{Cu } K\alpha) = 1.5418$ Å, $\mu = 18.07$ cm⁻¹, $F(000) = 640$, $T = 295$ K. Final $R = 0.050$ for 1708 observed reflections. The central ring of the tricyclic ring system is in a boat conformation. The folding angle between the planes of the pyrido and benzo rings is 158.7 (1)°. The plane of the 10-phenyl ring, almost perpendicular to the plane of the central ring [dihedral angle 95.4 (1)°], nearly bisects the tricyclic ring system.

Introduction. During the continuous search for biologically active phenothiazines, a series of substituted phenothiazines (Chiou, Reeves & Biehl, 1976; Biehl, Patrizi, Lapis & Reeves, 1974), heteroarylphenothiazines (Jovanovic & Biehl, 1983; Jovanovic, Biehl, Rosenstein & Chu, 1984), and their pyridobenzo-thiazine analogs (Jovanovic & Biehl, 1983) have been prepared. One of the interesting findings of these studies was the significant electronic interactions between the phenothiazine tricyclic ring and electron-withdrawing 10-aryl substituents. The examination of the interaction in the solid state by X-ray analysis can be used to compare the study of the same effects in solution by NMR spectroscopy (Jovanovic & Biehl, 1983, 1984).

* To whom correspondence should be addressed.

The *ortho*-substituted 10-arylphenothiazines have been used as the model compounds for assessing the relationship between resonance interactions as a function of the torsion angle between the two ring systems (Jovanovic & Biehl, 1984; Chu, Yang & van der Helm, 1976). The structure of 10-(2-methoxyphenyl)pyrido[3,2-*b*][1,4]benzothiazine (I) is reported in this paper as a reference compound for similar studies of 10-aryl-1-azaphenothiazines.



- (I) X = N, Y = OCH₃
 (II) X = CH, Y = OCH₃
 (III) X = N, Y = H

Experimental. Colorless, flat needles of (I) recrystallized from a solution of 15% CH₂Cl₂–85% absolute ethanol, unit-cell parameters by least-squares analysis of 15 reflections with 2θ values measured on the diffractometer, $P2_1/c$ deduced from systematic absences ($h0l$ absent with l odd, $0k0$ absent with k odd), Syntex $P2_1$ automatic diffractometer, crystal $0.43 \times 0.37 \times 0.07$ mm, $\theta/2\theta$ scanning mode, graphite-monochromated Cu $K\alpha$ radiation, 2529 independent reflections with $2\theta < 130^\circ$, range of hkl : $h \pm 13, k 0 \rightarrow 9, l 0 \rightarrow 18$, 1708 observed, $I > 3\sigma(I)$; three standard reflections measured repeatedly after every 50 reflections showed a random variation of less than 4% in intensity; Lorentz–polarization corrections, no absorption or extinction corrections; direct methods (*MULTAN78*, Main, Hull, Lessinger, Germain, Declercq & Woolfson, 1978), refinement by full-matrix least squares using *SHELX76* (Sheldrick, 1976), anisotropic, H positions located in successive difference Fourier syntheses, isotropic temperature factors for H atoms, $w = 1/[\sigma(F)]^2$, $\sigma(F)$ from counting statistics, $\sum w(|F_o| - |F_c|)^2$ minimized, $R = 0.050$, $R_w = 0.062$, $(\Delta/\sigma)_{\max} = 0.06$, max. height in final difference Fourier synthesis $0.21 \text{ e } \text{Å}^{-3}$, atomic scattering factors for S, O, N, C and H from *International Tables for X-ray Crystallography* (1974).

Discussion. Final parameters are given in Table 1.* The identification of the atoms and the configuration of the molecule are shown in the *ORTEP* (Johnson, 1965) drawing in Fig. 1. The bond lengths and bond angles, with their standard deviations, are shown in Fig. 2. The bond lengths and bond angles of the title compound are in good agreement with those of 10-(*o*-methoxyphenyl)phenothiazine (II) (Chu *et al.*, 1976).

* Lists of structure factors and anisotropic thermal parameters have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 39564 (15 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

Table 1. Fractional atomic coordinates ($\times 10^4$ for non-H, $\times 10^3$ for H atoms) with equivalent isotropic thermal parameters for non-H ($\times 10^4$) and isotropic for H atoms ($\times 10^3$) (e.s.d.'s are in parentheses)

$$U_{eq} = \frac{1}{3}(U_{11} + U_{22} + U_{33}).$$

	x	y	z	$U_{eq}/U(\text{Å}^2)$
C(1)	2017 (3)	79 (4)	8858 (2)	580 (8)*
C(2)	1183 (4)	616 (5)	9275 (3)	821 (11)
C(3)	221 (4)	1419 (5)	8844 (3)	849 (11)
C(4)	99 (3)	1683 (5)	7988 (3)	739 (12)
S(5)	827 (1)	1784 (1)	6486 (1)	750 (2)
C(6)	1534 (3)	-88 (5)	5283 (2)	634 (9)
C(7)	2260 (4)	-1144 (5)	4990 (2)	741 (10)
C(8)	3152 (4)	-1811 (5)	5564 (2)	720 (10)
N(9)	3319 (3)	-1528 (4)	6419 (2)	686 (7)
N(10)	2784 (2)	-220 (3)	7592 (2)	502 (6)
C(4a)	949 (3)	1194 (4)	7562 (2)	562 (8)
C(5a)	1690 (3)	274 (4)	6141 (2)	547 (7)
C(9a)	2579 (2)	-500 (4)	6710 (2)	485 (7)
C(10a)	1908 (2)	346 (4)	8007 (2)	496 (7)
C(1')	3764 (2)	-1014 (4)	8111 (2)	504 (7)
C(2')	3738 (3)	-2700 (4)	8278 (2)	525 (7)
C(3')	4690 (3)	-3441 (5)	8775 (2)	665 (9)
C(4')	5643 (3)	-2512 (6)	9097 (3)	761 (11)
C(5')	5680 (3)	-864 (6)	8930 (3)	817 (12)
C(6')	4729 (3)	-103 (5)	8426 (2)	685 (9)
O'	2748 (2)	-3504 (3)	7934 (1)	643 (6)
C(7')	2751 (6)	-5254 (6)	7935 (4)	932 (15)
H(1)	260 (3)	-50 (5)	916 (2)	114 (16)
H(2)	133 (3)	26 (5)	984 (2)	80 (12)
H(3)	-33 (5)	178 (6)	916 (3)	141 (19)
H(4)	-56 (3)	212 (5)	771 (2)	85 (12)
H(6)	95 (3)	37 (4)	493 (2)	72 (11)
H(7)	211 (3)	-145 (4)	439 (2)	72 (10)
H(8)	370 (4)	-249 (6)	537 (3)	94 (13)
H(3')	467 (3)	-449 (5)	886 (2)	67 (11)
H(4')	619 (4)	-299 (5)	945 (3)	91 (13)
H(5')	630 (4)	-9 (6)	911 (3)	110 (15)
H(6')	474 (3)	122 (5)	825 (2)	84 (11)
H(7')	205 (2)	-545 (8)	763 (4)	178 (27)
H(72')	288 (4)	-562 (7)	848 (3)	116 (19)
H(73')	329 (4)	-554 (5)	759 (3)	99 (15)

* The standard deviations of U_{eq} 's were calculated in accordance with the expressions derived by Schomaker & Marsh (1983).

The tricyclic ring system is folded with the central ring in a boat conformation as shown by the torsion angles in Fig. 2 and the puckering parameters (Cremer & Pople, 1975). The puckering parameters are $q_2 = 0.500$, $q_3 = 0.001$, $Q = 0.501 \text{ Å}$, $\varphi_2 = 4.0$ and $\theta = 101.0^\circ$. For ideal boat conformation, these values are $q_2 = Q$, $q_3 = 0 \text{ Å}$, $\varphi_2 = 0$ and $\theta = 90^\circ$. The 10-(2-methoxyphenyl) substituent is in a boat-equatorial conformation with respect to the central ring. The least-squares planes of the pyrido, benzo, and phenyl rings together with the deviations of atoms from the planes are shown in Fig. 1. The folding angle between the planes of the pyrido and benzo rings is $158.7(1)^\circ$ in (I) compared with the folding angle of 157.5° in (II). The phenyl ring nearly bisects the tricyclic ring system and the plane of the phenyl ring is nearly perpendicular to the plane of the central ring [C(4a), C(5a), C(9a) and C(10a)] with a dihedral angle of $95.4(1)^\circ$. The dihedral angles between the plane of the phenyl ring and the planes of the benzo and pyrido rings are $95.3(1)$ and $106.0(1)^\circ$ respectively. These data indicate that the substitution of the benzo ring in (II) by the pyrido ring does not have any significant effect on

o-methoxy group when compared with those of unsubstituted 10-phenylpyrido[3,2-*b*][1,4]benzothiazine (III). However, the largest differences observed are for the C atoms in the pyridine ring. This is supported by the conformation of the 10-(2-methoxyphenyl) substituent observed in the solid state. The *o*-methoxy group is tilted toward the pyrido ring as shown by the dihedral angles in Fig. 1 and the torsion angles about the N(10)–C(1') bond in Fig. 3. Furthermore, all of the *ortho* ¹³C–H couplings were too small to be observed in (III); however, the coupling constants in (I) have been observed and they are 3.10, 2.93, and 2.90 Hz, between C(8)–H(7), C(6)–H(7), and C(1)–H(2), respectively. These coupling changes indicate that there is an increase in the folding angle of the pyridobenzothiazine ring in (I) as compared to the unsubstituted 10-phenyl derivative (III). The crystal structure of (III) will be investigated to determine if the folding angle of (III) is smaller than that of (I) as predicted by NMR studies in solution.

The support of the Robert A. Welch Foundation to SSCC and ERB is gratefully acknowledged.

References

- BIEHL, E. R., PATRIZI, V., LAPIS, S. & REEVES, P. C. (1974). *J. Heterocycl. Chem.* **11**, 965–968.
- CHIOU, H., REEVES, P. C. & BIEHL, E. R. (1976). *J. Heterocycl. Chem.* **13**, 77–82.
- CHU, S. S. C., YANG, H. T. & VAN DER HELM, D. (1976). *Acta Cryst.* **B32**, 2567–2570.
- CREMER, D. & POPLE, J. A. (1975). *J. Am. Chem. Soc.* **97**, 1354–1358.
- International Tables for X-ray Crystallography* (1974). Vol. IV. Birmingham: Kynoch Press.
- JOHNSON, C. K. (1965). *ORTEP*. Report ORNL-3794. Oak Ridge National Laboratory, Tennessee.
- JOVANOVIĆ, M. V. & BIEHL, E. R. (1983). *J. Heterocycl. Chem.* **20**, 1677–1681.
- JOVANOVIĆ, M. V. & BIEHL, E. R. (1984). *Org. Magn. Reson.* In the press.
- JOVANOVIĆ, M. V., BIEHL, E. R., ROSENSTEIN, R. D. & CHU, S. S. C. (1984). *J. Heterocycl. Chem.* In the press.
- MAIN, P., HULL, S. E., LESSINGER, L., GERMAIN, G., DECLERQ, J.-P. & WOOLFSON, M. M. (1978). *MULTAN78. A System of Computer Programs for the Automatic Solution of Crystal Structures from X-ray Diffraction Data*. Univs. of York, England, and Louvain, Belgium.
- SCHOMAKER, V. & MARSH, R. E. (1983). *Acta Cryst.* **A39**, 819–820.
- SHELDRIK, G. M. (1976). *SHELX76*. Program for crystal structure determination. Univ. of Cambridge, England.

Acta Cryst. (1984). **C40**, 1756–1758

Thuriferin Monohydrate, C₂₁H₃₄O₄·H₂O

BY IVAN VICKOVIĆ* AND WILLIAM H. WATSON

FASTBIOS Laboratory, Department of Chemistry, Texas Christian University, Fort Worth, Texas 76129, USA

AND MARIO SILVA AND MAGALIS BITTNER

Laboratorio de Productos Naturales, Departamento de Botánica, Universidad de Concepción, Concepción, Chile

(Received 14 February 1984; accepted 30 May 1984)

Abstract. $M_r = 368.52$, monoclinic, $P2_1$, $a = 7.131$ (1), $b = 21.310$ (3), $c = 7.033$ (1) Å, $\beta = 110.93$ (1)°, $V = 998.2$ (3) Å³, $Z = 2$, $D_x = 1.226$ Mg m⁻³, $\lambda(\text{Cu } K\alpha) = 1.54178$ Å, $\mu = 0.697$ mm⁻¹, $F(000) = 404$, room temperature, $R = 0.046$ for 1257 unique reflections with $I \geq 2\sigma(I)$. Thuriferin is a kaurane-type diterpene isolated from *Robinsonia thurifera* (Mol.) D.C. (tribe Senecioneae of family Compositae). It contains three six-membered rings in chair conformations, a five-membered ring in an envelope conformation, and an unusual methyl hemiacetal functionality.

Introduction. The concept of the tribe Senecioneae (Bentham, 1873) was generally accepted and remained basically unchanged for almost a century. During recent years there has been some disagreement concerning the number of subtribes, and a number of genera have even been shifted from tribes such as Helenieae into an expanded concept of Senecioneae (Nordenstam, 1977). Others have narrowed the concept and have excluded anomalous genera and even the entire subtribe Liabinae (Turner & Powell, 1977). In an attempt to provide chemical data for use as additional taxonomic markers, we have investigated *Robinsonia thurifera*. The genus *Robinsonia* contains approximately six species of rosette trees or epiphytic plants on tree ferns.

* On leave from the University Computing Center, Zagreb, Yugoslavia.